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#### **Key indicators**

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.004 Å R factor = 0.063 wR factor = 0.150 Data-to-parameter ratio = 13.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title molecule,  $C_{15}H_8CINO_2$ , is essentially planar. The molecular packing in the crystal is stabilized by weak intermolecular  $C-H\cdots\pi$  interactions and van der Waals forces.

3-Chlorobenzoxazolo[3,2-b]isoquinolin-6-one

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#### Comment

We have recently investigated the photocycloaddition reactions with electron-deficient alkenes (Ling *et al.*, 1999) and photoinduced oxygenation reactions of a new class of polycyclic heterocycles, namely 11H-benzoxazolo[3,2-*b*]isoquinolin-11-ones, which are benzannelated 5H-oxazolo[3,2-*a*]pyridine derivatives. The title compound, (I), one of the derivatives, was isolated from the reaction mixture of the reaction of homophthalic acid with 2-amino-4-chlorophenol. An X-ray crystal structure determination of (I) was undertaken in order to elucidate the conformation, and the results are presented here.



The bond lengths and angles in (I) are in good agreement with those observed for a closely related structure (Ravishankar *et al.*, 2001). The molecule is essentially planar, with a maximum deviation of 0.043 (3) Å for atom C11. The dihedral angle between the planes of the isoquinoline and benzoxazole moieties is 1.46 (8)°. The coplanarity extends the  $\pi$ -conjugation between these two aromatic substituents. The crystal packing is stabilized by weak intermolecular C–H··· $\pi$  interactions, involving both aromatic rings of the isoquinoline moieties of glide-related molecules.

#### **Experimental**

The title compound was prepared by refluxing homophthalic anhydride (25 mmol) and 2-amino-4-chlorophenol (25 mmol) in acetic acid for 6 h. The solid product was collected by filtration. Single crystals suitable for crystallographic analysis were obtained by slow evaporation of an acetone-petroleum ether (1:20 v/v) solution.

#### Crystal data

C <sub>15</sub> H <sub>8</sub> ClNO <sub>2</sub>	$D_x = 1.550 \text{ Mg m}^{-3}$
$M_r = 269.67$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 2228
a = 23.831 (2)  Å	reflections
b = 5.9716 (6) Å	$\theta = 2.6-28.3^{\circ}$
c = 8.1549 (8)  Å	$\mu = 0.33 \text{ mm}^{-1}$
$\beta = 95.195 \ (2)^{\circ}$	T = 293 (2) K
$V = 1155.7 (2) \text{ Å}^3$	Block, colorless
Z = 4	$0.40 \times 0.24 \times 0.24$ mm

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# organic papers

#### Data collection

Siemens SMART CCD area- detector diffractometer	2823 indeper 1976 reflection
$\omega$ scans	$R_{\rm int} = 0.033$
Absorption correction: multi-scan	$\theta_{\rm max} = 28.3^{\circ}$
(SADABS; Sheldrick, 1996)	$h = -31 \rightarrow 1$
$T_{\min} = 0.881, T_{\max} = 0.926$	$k = -7 \rightarrow 7$
7010 measured reflections	$l = -10 \rightarrow 1$
Refinement	
Refinement on $F^2$	$w = 1/[\sigma^2(F_o)]$

Remember on F  $R[F^2 > 2\sigma(F^2)] = 0.063$   $wR(F^2) = 0.150$  S = 1.082823 reflections 204 parameters All H-atom parameters refined 823 independent reflections 976 reflections with  $I > 2\sigma(I)$   $R_{int} = 0.033$   $max = 28.3^{\circ}$   $z = -31 \rightarrow 23$   $z = -7 \rightarrow 7$  $z = -10 \rightarrow 10$ 

$$\begin{split} w &= 1/[\sigma^2(F_o^2) + (0.0601P)^2 \\ &+ 0.7591P] \\ \text{where } P &= (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\max} &< 0.001 \\ \Delta\rho_{\max} &= 0.26 \text{ e} \text{ Å}^{-3} \\ \Delta\rho_{\min} &= -0.40 \text{ e} \text{ Å}^{-3} \end{split}$$

#### Table 1

Selected geometric parameters (Å, °).

1.742 (3)	C1-N1	1.400 (3)
1.367 (3)	C7-C8	1.334 (4)
1.381 (3)	C7-N1	1.384 (3)
1.216 (3)	C15-N1	1.401 (3)
-178.5 (2)	O1-C7-C8-C9	179.8 (2)
	1.742 (3) 1.367 (3) 1.381 (3) 1.216 (3) -178.5 (2)	$\begin{array}{cccc} 1.742 (3) & C1-N1 \\ 1.367 (3) & C7-C8 \\ 1.381 (3) & C7-N1 \\ 1.216 (3) & C15-N1 \\ \end{array}$

#### Table 2

Hydrogen-bonding geometry (Å, °).

Cg(A) and Cg(B) denote the centroids of the pyridine and benzene rings of the isoquinoline moiety.

D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
0.92 (3) 0.96 (3)	2.79 (3) 2.75 (3)	3.532 (4) 3.476 (4)	139 (2) 133 (2)
	<i>D</i> -H 0.92 (3) 0.96 (3)	$\begin{array}{c c} D-H & H\cdots A \\ \hline 0.92 (3) & 2.79 (3) \\ 0.96 (3) & 2.75 (3) \end{array}$	$D-H$ $H\cdots A$ $D\cdots A$ 0.92 (3)         2.79 (3)         3.532 (4)           0.96 (3)         2.75 (3)         3.476 (4)

Symmetry codes: (i)  $x, -\frac{1}{2} - y, \frac{1}{2} + z$ ; (ii)  $x, \frac{1}{2} - y, z - \frac{1}{2}$ .

All H atoms were located from a difference Fourier map and were refined isotropically. The C-H distances are in the range 0.89 (3)–0.96 (3) Å.



#### Figure 1

The structure of the title compound, showing 50% probability displacement ellipsoids and the atom-numbering scheme.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

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